

## Features of the Diels-Alder reaction between 9,10-diphenylanthracene and 4-phenyl-1,2,4-triazoline-3,5-dione

Kiselev V., Kornilov D., Kashaeva E., Potapova L., Krivolapov D., Litvinov I., Konovalov A.  
*Kazan Federal University, 420008, Kremlevskaya 18, Kazan, Russia*

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### Abstract

© 2014 Pleiades Publishing, Ltd. The Diels-Alder reaction between substituted anthracenes 1a-1j and 4-phenyl-1,2,4-triazoline-3,5 (2) is studied. In all cases except one, the reaction proceeds on the most active 9,10-atoms of substituted anthracenes. The orthogonality of the two phenyl groups at the 9,10-position of diene 1a is found to shield 9,10-reactive centers. No dienophiles with C=C bonds are shown to participate in the Diels-Alder reaction with 1a; however, the reaction 1a + 2 proceeds with the very active dienophile 2,4-phenyl-1,2,4-triazoline-3,5-dione. It is shown that attachment occurs on the less active but sterically accessible 1,4-reactive center of diene 1a. The structure of adduct 3a is proved by  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectroscopy and X-ray diffraction analysis. The following parameters are obtained for reaction 1a + 2  $\rightleftharpoons$  3a in toluene at 25°C:  $K_{\text{eq}} = 2120 \text{ M}^{-1}$ ,  $\Delta H_f^\circ = 58.6 \text{ kJ/mol}$ ,  $\Delta S_f^\circ = -97 \text{ J/(mol K)}$ ,  $\Delta V_f^\circ = -17.2 \text{ cm}^3/\text{mol}$ ,  $\Delta H_b^\circ = 108.8 \text{ kJ/mol}$ ,  $\Delta S_b^\circ = 7.3 \text{ J/(mol K)}$ ,  $\Delta V_b^\circ = -0.8 \text{ cm}^3/\text{mol}$ ,  $\Delta H_{r-n} = -50.2 \text{ kJ/mol}$ ,  $\Delta S_{r-n} = -104.3 \text{ J/(mol K)}$ ,  $\Delta V_{r-n} = -15.6 \text{ cm}^3/\text{mol}$ . It is concluded that the values of equilibrium constants of the reactions 1a-1j + 2  $\rightleftharpoons$  3a-3j vary within  $4 \times 10^1$ - $10^{11} \text{ M}^{-1}$ .

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### Keywords

4-phenyl-1,2,4-triazoline-3,5-dione, 9,10-diphenylanthracene, Diels-Alder reaction, high pressure, regioselectivity